

Application of chlorhexidine and its effect on hybrid layers: A review of the literature

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ABSTRACT

Background: Adhesion must occur between the demineralized organic mass of the tooth structure and the hydrophilic elements of the adhesive resin. The goal of achieving clinical success with this procedure is clearly expressed in the recommendations for the application of the acidifying and bonding materials, as well as in the rules that appear under the dental material manufacturing brand. Bonding the composite with demineralized dental structures is a procedural technique that relies on the sensitive mechanism of adhesion between the hydrophobic and hydrophilic parts of the resin and the dental organic matrix after acidification. The clinical success of the application is then expressed through the longevity of the filling and the absence of secondary or subsequent complications. **Materials and Methods:** This study is a literature review. A total of 54 articles were selected, of which 12 were excluded from further evaluation as they did not meet one or more of the inclusion criteria in the study. Therefore, 42 articles were included as the base articles for this study. **Results:** Chlorhexidine (CHX) was applied after rinsing the acid with water and air-drying the dentin only on a wet surface. This method is the only clinically proven one, as it is easy to apply and likely to gain wider acceptance initially. However, research is expected to develop simpler and more efficient methods of applying CHX or other matrix metalloproteinase inhibitors. For instance, a recent *in vitro* study demonstrated that the addition of 2% CHX to a conventional 37% phosphoric acid prevented the degradation of bond strength in an “Etch and Rinse” adhesive system over a 6-month period. **Conclusion:** The clinical application of 2% CHX for 1 min on acidified dentin, following an acid rinse and before the application of dentin primer and bond, has been shown to effectively prevent significant *in vivo* degradation of bond strength by metalloproteinases for a minimum of 14 months.

Key words: Chlorhexidine, Degradation, Hybrid layer

The effect of the concentration and time of application of chlorhexidine (CHX) has been evaluated, showing that times <1 min and concentrations <2% can prevent degradation *in vitro*. However, these results have not yet been verified *in vivo*. These findings suggest that the success of adding CHX to self-acidification systems to prevent hybrid bond degradation will vary with the resins used in commercial products. *In vitro* application of concentrations >0.12% CHX to dentin before the primer application is contraindicated, as it may reduce 24-h bond strength [1-7].

The application of this solution, either in its composition or as part of the primer, highlights the fact that CHX is used in different concentrations for various treatment purposes. This is

the element that attracts attention, with the ups and downs in the appearance of the advantages and disadvantages of applying this solution. Furthermore, the theoretical support for the fixation of composite in tooth structures lies in microretention. Regardless of the principles of tooth preparation, tooth structures, such as dentin or enamel, exhibit different hardness and relief roughness. In addition to these characteristics, dentin and demineralized enamel contain numerous “tunnels” and interprismatic spaces and dentinal tubules in quantity, with different numbers per surface unit. The penetration of the adhesive resin into the tubules, opened by acidification, enhances the adhesive surface of the resin to the tooth structure [7-10].

The structure, inorganic content, and organic content of both enamel and dentin undergo specific changes during the initial acidification process and subsequent bonding. The key to the whole repair process with the dental restoration in the tooth lies

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in the hybrid layer or the transition layer, from the polymerized adhesive resin to the demineralized layers of the enamel and dentin. This layer plays a key role in maintaining the integrity of the tissues and in preventing unnecessary structural degradation caused by the activation of enzymes of dissolved collagen [11-13]. A review of the literature about the elements reveals various substances used to inhibit the degradation of the hybrid layer, each differing in their chemical characteristics, such as acid or basic orientation and organic structural composition.

Glutaraldehyde, also known by its trade name Gluma, is one such element applied to prevent degradation of the hybrid layer between dentin and the adhesive material [8,11,14]. Several studies have examined the quantitative and qualitative effects of glutaraldehyde on tooth weight loss, if any, or the release of hydroxyapatite [14-20]. These effects have been assessed at different time points during the clinical lifespan of the completed filling. Therefore, to clarify whether degradation of the layer occurs immediately on acidification or is a time-dependent process, it is important to maintain the stability of collagen fibrils before dentin treatment with polyphenols. The application of CHX is one such method [21-25].

The strategy of dentinal biomodification with proanthocyanidin extracts [21-29] is another source of literature that is oriented around the application of preservation solutions against dentinal degradation. Another strategy is the application of extrafibrillar antimicrobial agents, with kyotosan being one of them, to preserve intrafibrillar minerals within intact collagen fibrils and protease activation. The use of 1% zinc oxide (ZnO) can ensure the integrity of the hybrid layer, reduce cytotoxicity, and minimize the dissolution of the dentinal adhesive [11,30-37].

Factors that influence the study of this connection can include: [24-34]

- Physical forces of displacement between two solid structures connected by microretention, resulting from occlusal forces. These schooling forces can also express the action as the horizontal component of the occlusal force, applied within the coping norms of the forces on the part of the dental structure
- Given that the hybrid layer consists of two structures with different dissolution temperatures or undergoing physical shape changes, it is understood that when different temperatures are applied in the connecting contour between the resin and dentine, the reaction of the two-component structures of the hybrid layer is different
- Substances with different acids or basic contents can penetrate the hybrid layer, which can react and cause the dissolution of the structure between the dentin and the adhesive resin
- During acidification and bonding, the work protocols, and whether or not moisture is properly maintained, allow excess water in the dentinal fluid. This excess water acts as an initiator for the degradation of the hybrid layer
- Class I dentinal collagen is broken down by endogenous metalloproteinases or by those of bacterial origin, whose activation occurs and is not hindered by the degradation process of the hybrid layer

- Different resins have different modes and factors of degradation [21-25].

Moreover, if degradation of the hybrid layer occurs, the stimulus comes from the activation of endogenous metalloproteinases inside the dentinal structure, which destroys the microretention of the filling realized with the composite. The larger the surface to be acidified or bonded, the larger the surface area of the hybrid layer created, and the higher the microstress created by the filling [22,23-29].

MATERIALS AND METHODS

The study focuses on collecting already published data on the effect of CHX on inhibiting hybrid layer degradation. The application of this solution, either simple in composition or as part of a primer, draws attention to the fact that CHX solution is used in different concentrations and for different treatment purposes. This is the element that attracts attention with the ups and downs in the appearance of the advantages and disadvantages of applying this solution [1,3,30-38].

The collection of information should be oriented around finding the trend of scientific research in this regard, toward how this data is required through the results according to studies of the review type, or through studies based on experiments performed *in vitro* or *in vivo* on permanent teeth during composite restorations [1,38-42].

The conducted study is a literature review. An electronic search at a time search interval of 15 years was performed on PubMed using the keywords: Hybrid layer and CHX and multistep degradation in combination of words [43-78]. After analyzing the collected abstracts and articles, the inclusion and exclusion criteria were applied. The inclusion criteria in the analysis were all the articles that directly evaluated the impact of dentinal biomodification agents on the degradation of the hybrid layer. The exclusion criteria were as follows:

1. Studies that do not directly mention both keywords simultaneously: Adaptation of the hybrid
2. Layer and degradation of the hybrid layer
3. Studies aimed at the strength of the adhesive layer, which had no influence or connection with the hybrid layer
4. Studies that discussed acidification and bonding of glass-ionomers

A total of 54 articles and 18 books were initially selected, of which 12 articles were excluded from further evaluation as they did not meet one or some of the following criteria:

1. Articles that do not mention periodontal diseases
2. Articles that discuss periodontal diseases, but do not classify them
3. Articles that do not discuss the treatment of each periodontal disease
4. Articles that discuss not-yet-approved periodontal card systems.
5. At this stage, 42 articles were selected as part of the basic articles included in the study [43-78]

RESULTS

The goal of achieving clinical success in this procedure is clearly expressed in the recommendations for the application of the acidifying and bonding materials, and the rules provided by the brand of dental material fabrication. The adhesion of the composite to demineralized dental structures is a technical procedure that relies on the sensitive mechanism of adhesion of the hydrophobic and hydrophilic parts of the resin to the organic dentinal matrix after acidification. The clinical success of the application is demonstrated by the longevity of the restorations and the absence of secondary or subsequent complications [1,37-42,79-85]. At the microscopic level, this discussion boils down to analyzing what happens in the hybrid filling layer, which is the initial contact of the adhesive resin with the dentinal tubules or the interprismatic spaces. This contact, depending on the conditions and the level of humidity tolerated by the adhesive resin, following the manufacturer's recommendations can either remain intact or begin to degrade as an intermediate layer, which will further reflect its effect on the lifespan clinic of the existing filling [1,79-82].

Bonding as a procedure heavily relies on the method of applying the adhesive layer, whether it is through active or passive application methods. This is primarily oriented by the presence or absence of moisture on the contact surfaces with the demineralized areas. Another element that attracts attention in this second stage of tooth filling is the thickness of the adhesive resin layer in the composite filling. This thickness primarily depends on the thickness of the hybrid layer on the demineralized dentinal structural surfaces, a thickness that is oriented by the penetrating and active ability of the selected acidification type. Table 1 presents all the articles collected to understand the trend of scientific research over the years about CHX as an inhibitory solution for matrix metalloproteinase (MMPs).

Another element has to do with the technical procedure of applying the adhesive resin, which means how much we apply on the demineralized surface of the tooth structure. Here, we intervene to emphasize some recommendations for the application of the air current to orient the mechanical dispersion of the bond on the demineralized dentinal surface of the tooth prepared for filling with composite.

The thickness of the hybrid layer, based on the literature, is highly dependent on the penetration of acidification and the

barrier of the adhesive resin against the demineralized dentinal tubules. If we look at this procedure on a microscopic level, it is also related to the process of hydrophobic and hydrophilic interconnections and interactions of the mechanical passage of the adhesive resin against tubules of different dentin sizes. After this transition, the tendency carried out by the force exerted by the air pressure above the non-polymerized layer of the resin adhesive makes another mechanical distribution of this layer against the partially filled dentinal tubules with collagen fibrils, which can even be partially wet [43,44,83,84].

Adhesive materials and their application must be strictly implemented according to the manufacturer's recommendations. These recommendations, which pertain to the time of application and the technical complexity, are the factors that influence the modification of these materials in specific patients or pediatric patients. These modifications aim to streamline the application process and potentially reduce the overall application time. The reduction of the clinical application time, by reducing a step of the filling with composite, is another advantage of these adhesive materials. Another important element is whether dryness is preserved during the application of the composite filling material. This involves considering whether the material is applied in the presence of moisture or on completely dry demineralized surfaces, that is, achieving a chalk-like appearance on enamel or a whiter color on demineralized dentinal structures compared to neighboring structures. The method of applying adhesive materials also differs based on the following classification: Active and passive application. The difference between these two methods lies in the process of rubbing the material against the dental structures, which is thought to increase the strength of the bond to the hybrid layer. Finally, the micropores caused by demineralization play a crucial role in facilitating the interdiffusion between the enamel and the dentine hybrid layer [43,44,80-85].

DISCUSSION

Based on the sources found in the literature, it can be stated that further studies are needed to assess the degradation of the hybrid layer and obtain long-term clinical data [1,34-42,79,80]. The link between these factors is directly related to the aggressiveness of the acid action, which is expressed in the duration of the acid

Table 1: All the articles collected to see the trend of scientific research over the years about chlorhexidine as an inhibitory solution of MMPs

Study - Author	Aim	Time	Results
Favetti et al. 2017 [77]	2% CHX digluconate cervical lesions clinical evaluation	36 months 1 week, 6 months, 12, 24, 36 months	There are no differences As in the case of application or non-application of CHX 2%
Brackett et al. 2007 [70]	TEM - Class I 2% CHX gluconate Microtensile strength of the bond	2-6 months	CHX 2% digluconate is effective in maintaining the hybrid layer
Hebling et al. 2005 [52]	TEM assessment CHX class I Black	6 months	CHX maintains normal stability in the integrity of the hybrid layer
Carrilho et al. 2007 [53]	CHX after phosphoric acid Class I Microtensile strength of the bond	14 months	Maintains the stability of the integrity of the CHX hybrid layer
Brackett et al. 2009 [65]	Class I 2% of CHX digluconate Microtensile strength of the bond	12 months	CHX 2% of effective digluconate

MMPs: Matrix metalloproteinases

application and the percentage of acidity. In addition, the depth of the hybrid layer is directly proportional. Consequently, a reduction of the hybrid layer leads to decreased adhesive infiltration and leads to the appearance of fewer partially demineralized areas that are susceptible to hybrid layer degradation [13-17]. Although the demineralizing morphology may be similar for certain types of acidifications, the possibility of creating adaptations of the hybrid layer in the filling and preventing its degradation should be considered.

Depending on the type of acidification, the creation and adaptation of the hybrid layer against the adhesive resin and the dental structure where the filling will be performed exhibit different characteristics [2-5]. The degradation or dissolution of the hybrid layer should be viewed as the process of the collapse of the fiber structure, specifically Type I collagen, that remains after acidification. Moreover, it is understood that the degradation of this structure must be seen from the perspective of the activation of collagen-destroying tissue enzymes found in the dentinal tissue matrix. Metalloproteinases and cathepsin cysteine are the enzymes responsible for tissue degradation. When examining the organic content of dentin, it is observed that approximately 90% of the organic part consists of Type I collagen, while the remainder comprises non-collagenous proteins. Among these proteins are the group of metalloproteinases, which are thought to be responsible for the degradation of the hybrid layer. These enzymes are sensitive to the presence of certain ions, such as dentinal Zn and Ca. These enzymes are in an inactive form and can be activated locally by cleaving the dimer with the catalytic zinc [3-7].

Another aspect of hybrid layer degradation should be considered in terms of the hydrolysis of the adhesive resin, which can be initiated as a result of the application error of the adhesive layer. This is observed more frequently in adhesive systems with multiple application stages compared to systems with a single application [3-7]. Technological advancements aim to prevent the degradation of the hybrid layer, as it gradually leads to the development of complications in composite fillings [7]. The degradation of the hybrid layer occurs as a result of the hydrolysis of the adhesive resin and the dissolution of collagen, both components of the hybrid layer. The hydrolytic stability of the hybrid layer differs from the application technique of the adhesive layer, thereby expressing higher results in systems with different stages of application than in systems with united stages of application. The breakdown of collagen and the release of collagen Type 1 occurs due to the activation of metalloproteinases and cysteine cathepsin, which are responsible for the hydrolytic degradation of the collagenous part of the hybrid layer. After the demineralization of the dentin during acidification, the moisture of the rinsing process takes place in the preservation of the collagenous structure. In addition, the resin cannot remove water residues but leaves behind layers of demineralization residues, which fall prey to the hydrolytic degradation of the hybrid layer, after the activation of metalloproteinases.

Santos *et al.* considered that 2% of CHX, either in aqueous solution or gel form, as a non-oxidizing agent, did not interfere

with the interaction of an adhesive system in pulp chamber dentin. However, an exception to this trend was observed when CHX gel was combined with ethylenediaminetetraacetic acid (EDTA). Although the base of the gel used with CHX is a water-soluble carbon polymer, which can be easily removed from the root canal, an occasional presence of residual gel in the dentin may react with EDTA. This reaction can lead to the formation of products that affect resin infiltration and/or polymerization, resulting in slightly lower bond strength values compared to the control group. Therefore, it is important to make every effort to remove traces of chemical substances from the canal through intermediate rinses with inert solutions [43,44,83,84]. CHX has demonstrated the ability to maintain hybrid layer stability and bond strength in both *vitro* and *in vivo* studies, possibly due to its effectiveness as an MMP inhibitor. This inhibitory effect results in reduced degradation of the hybrid layer and collagen fibrils. It is a remarkable property because one of the reasons for the long-term loss of integrity in composite-dentin bonds is the degradation of collagen fibrils exposed in incomplete infiltrated hybrid layers. This degradation is attributed to an endogenous proteolytic mechanism involving the activity of MPMs present in dentin [20-22].

CONCLUSION

The application of CHX and its significance in preventing the degradation of the hybrid layer has been confirmed in the published sources found in the literature. The method of its application is easy as an application procedure. In the future, adhesive systems that allow the gradual release of CHX over an extended period will be developed. According to the literature, the effects of CHX can be sustained even up to 10 years after application. However, it is worth noting that there are relatively few literature sources and studies in this field, and they are primarily focused on *in vitro* research using extracted teeth rather than *in vivo* studies on natural teeth.

AUTHORS' CONTRIBUTION

All authors have made a significant contribution to this work, be it in the conception, implementation, and literature review or in all these areas; took part in drafting, revising, or critically reviewing the article; gave final approval of the version to be published; have agreed on the journal to which the article has been submitted; and agree to be responsible for all aspects of the work.

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